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Highly Regioselective Radical Cyclizations of Allenamides

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ABSTRACT

The first radical cyclizations of allenamides are described. These reactions are highly regioselective for the *central* carbon of the allenic moiety, leading to an efficient preparation of nitrogen heterocycles such as isoquinolines, and carbocycles such as indane and naphthalene derivatives. The *exo*-cyclization mode could also be achieved in some cases, leading to the synthesis of isoindoles. The feasibility of a tandem radical cyclization using allenamide is also established.

Radical reactions represent a powerful tool in organic synthesis.^{1–4} Our efforts in developing synthetic methods employing allenamides^{5–7} led us to speculate on the possibility of a radical cyclization using allenamides **1** [Figure 1]. In addition to achieving a new approach for constructing nitrogen heterocycles [see **4**, **6**, or **8**], this investigation presents a unique opportunity to address the regioselectivity issue in radical processes involving allenes: *endo*- versus *central*- versus *exo*-cyclization, which is not a well-resolved problem.^{8–11}

While both *endo-* and *exo-*cyclizations of radical intermediate 2 would lead to new vinyl radicals 3 and 7,

respectively, *central*-cyclization would give the energetically more favored allyl radical **5** [Figure 1]. Since this is an intramolecular radical process, *endo*-cyclization may be prohibited due to geometric constraints. With these issues in mind, we investigated radical cyclizations of allenamides. We report here a highly regioselective radical cyclization of allenamides for syntheses of isoquinolines and isoindoles.

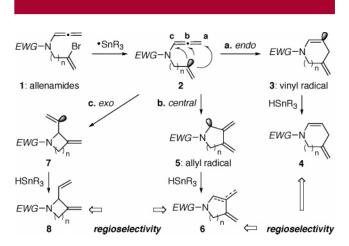


Figure 1.

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Table 1.

entry	SM	initiator [equiv]	H-donor [equiv]	temp (°C)	$\mathrm{solvent}^a$	product	$yield^b$
1	12a	AIBN [0.4]	<i>n</i> -Bu ₃ SnH [2.0]	80	PhH		no rxn ^c
2	12a	AIBN [0.4]	n-Bu ₃ SnH [2.0]	110	tol		no rxn^c
3	12a	Et ₃ B/air	TMS_3SiH [1.1]	50	tol		decomp
4	12a	Et ₃ B/air	TMS_3SiH [2.0]	50	tol		decomp
5	12a	$[PhCO]_2O_2$ [0.2]	TMS_3SiH [1.1]	50	tol		decomp
6	12b	AIBN [0.4]	n -Bu $_3$ SnH [1.5]	80	tol	14	66%
7	12b	AIBN[0.4]	n-Bu ₃ SnH [1.5]	80	tol^d	14	44
8	12b	AIBN[0.4]	TMS_3SiH [1.1]	80	tol	14	10
9	12b	AIBN[0.4]	PhSH	80	tol		no rxn^e
10	12b	AIBN[0.4]	$t ext{-BuSH}$	80	tol		no rxn^e
11	12b	Et ₃ B/air	n-Bu ₃ SnH [2.0]	rt	tol	14	10
12	12b	Et ₃ B/air	TMS_3SiH [1.1]	rt	tol	14	8
13	12b	$[PhCO]_2O_2[0.2]$	n-Bu ₃ SnH [2.0]	60	tol	14	44
14	12b	$[PhCO]_2O_2[0.2]$	n-Bu ₃ SnH [2.0]	60	told	14	48

^a Concentration is 0.05 M unless otherwise noted. ^b Isolated yield. ^c Partial recovery of **12a** and decompositions. ^d Concentration = 0.01 M. ^e Addition of thiols to allene were found in high yields.

To establish the feasibility, allenamides **12a** and **12b**¹² with either a bromo- or iodo-benzyl group substituted at the

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nitrogen atom [*N*-tethered] were prepared from aryl halides **9a** and **9b**, respectively, in good overall yields [Scheme 1]. This preparation features base-induced isomerization, ¹³ which still represents the most reliable and facile synthetic entry to various allenamides.

Given the lack of precedent in radical cyclizations using allenamides, *N*-tethered allenamides **12a** and **12b** were employed in the screening of a range of conditions. As summarized in Table 1, it appears that only iodide is a suitable radical precursor as only the iodo-benzyl substituted allenamide **12b** underwent cyclizations [entries 6–14],

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⁽¹²⁾ All new compounds were characterized by 1H NMR, ^{13}C NMR, FTIR, $[\alpha]^{20}_D,$ and MS; see Supporting Information.

whereas **12a** was not useful [entries 1–5]. In addition, although AIBN is a better initiator than benzoyl peroxide [entries 6 and 7 versus 13 and 14], *n*-Bu₃SnH appears to be the best hydrogen donor, with optimal reaction temperature being 80 °C and concentration at 0.05 M [entry 6 versus 7]. Most significantly, only isoquinoline **14** was isolated as a consequence of a highly regioselective radical cyclization onto the *central* carbon **b**. Neither the *endo*-cyclized product isobenzazepin **13** nor the *exo*-cyclized product isoindole **15** was found.

This specific regioselectivity is further displayed using a range of different allenamides **16a**–**f** that contains a urethane [entries 1 and 2], urea [entry 3], or amido substitution [entries 4–6] [Table 2]. These results suggest that the electronic

Table 2.

entry	allenamides	product	$\operatorname{yield}^{a}\left(\%\right)$
1	16a : $R = OtBu$	17a	75
2	16b : $R = O-(+)$ -menthyl	17b	80
3	16c : $R = NMe_2$	17c	69
4	16d : $R = Me$	17d	58
5	16e : $R = [CH_2]_2CH = CH_2$	17e	55^b
6	16f : $R = isopropenyl$	17f	44^{c}

 a Isolated yield. b 42% and 46% yields at 0.01 and 0.005 M, respectively. c 44% yield at 0.01 M.

nature of the allenamide nitrogen atom has no impact on the regioselectivity. In addition, cyclization of allenamide 18 substituted with an iodo-phenyl group [one less methylene unit than 12b] was also feasible in favor of *central*cyclization, providing indole 19 in 91% yield [Scheme 2].

We next examined allenamides **20** and **21** with the aryliodide group tethered through the α -carbon of the allenamide [C-tethered] Scheme 3]. Their preparations involved the α -lithiation and alkylation of the corresponding unsubstituted allenamide. ¹⁵ Employing similar reaction conditions, the

Scheme 3

0.40 equiv AIBN
1.5 equiv n-Bu₃SnH

toluene, 0.05 M, 80 °C

Yield: 70%

20: C-tethered

22: central

Ph

Ph

Ph

21: C-tethered

23: central [76%]

24: exo [0%]

radical cyclization of **20** led to indene **22** in 70% yield as the exclusive product. Allenamide **21** with an extra methylene unit [see the box] was also suitable for the radical cyclization and afforded only the *central*-cyclized product dihydronaphthalene **23**. The *exo*-cyclized Indane product **24** was not found

It was surprising, then, to find that reactions of N-tethered allenamides **25** and **26**, containing a carbonyl group [see the box] at the tethering, did give *exo*-cyclized products [Scheme 4]. In fact, the *exo*-cyclization product **27** was the major

product. Although the overall yield is a modest 35%, this establishes the possibility of allenamide radical cyclization in an *exo*-mode.

The use of allenamide **26** containing a chiral auxiliary further demonstrates the feasibility of an *exo*-cyclization and

(14) **Typical Experimental Procedure.** A solution of allenamide **16a** (175.5 mg, 0.47 mmol) and AIBN (0.4 equiv, 30.8 mg) in toluene (9.4 mL) was heated to 80 °C for 5 min, after which, n-Bu₃SnH (1.5 equiv, 189.5 μ L) was added dropwise via a syringe. [Note: a syringe pump was used for the tandem reaction.] The reaction was heated at 80 °C for 3 h before being cooled to room temperature. A saturated aqueous KF solution (5 mL) was then added, and the mixture was stirred for another 2 h at room temperature. After filtration through Celite, the biphasic filtrate was extracted with 5 mL of EtOAc. The aqueous layer was extracted with another 5 mL portion of EtOAc after separation. The combined organic extracts were concentrated under reduced pressure, and the crude residue was purified using flash silica gel column chromatography (gradient eluent: 0–10% EtOAc in hexane) to provide 86.1 mg (75% yield) of compound **17a** as thick yellow oil.

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provided the *exo*-product **29** as a single diastereomer, 16 although the *exo* to *central* ratio dropped to 1.6:1. The sp² hybridized carbonyl carbon evidently provides sufficient constraint to allow a better *exo*-cyclization trajectory onto carbon **c**.

We were able to subsequently pursue a tandem radical cyclization^{4,8} using allenamide **31** as shown in Scheme 5.

Syringe pump addition of *n*-Bu₃SnH was needed for this reaction, and under these conditions, the desired tandem cyclization product **32** was isolated along with the arrested intermediate **33** in an overall 34% yield.¹⁷ Both products can

be derived from an initial *exo*-cyclization of allenamide **31**, likely through the rotamer **31a** via radical intermediates **34a** and **35**.

We also isolated the bridged heterocycle **36** in 17% yield [Scheme 5]. This is most likely a result of an initial radical cyclization onto not the allenic motif but the allyl group in rotamer **31b** given the favorable proximity, leading to a secondary radical intermediate **37** prior to cyclization onto the allene. We are currently exploring optimization of this radical cyclization pathway.¹⁸

Finally, it is noteworthy that an initial *exo*-cyclization onto the allenic motif [see **35**] appears to be prerequisite for completing the tandem process. In the case of allenamide **16e** [Scheme 5], although a tandem cyclization could have occurred under the same reaction conditions, leading to **40** after an initial *central*-cyclization, it was not observed. Instead, the allyl radical intermediate **39** was simply trapped to give the corresponding isoquinoline **17e** [see Table 2]. This difference can be attributed to the fact that the more stable allyl radical **39** was simply not as reactive, and thus the second cyclization was much slower relative to abstraction of hydrogen from n-Bu₃SnH.

We have described here the first radical cyclizations employing allenamides. These reactions are highly regioselective for the central carbon in the allenic motif, leading to a useful preparation of isoquinolines as well as derivatives of indole, indane, and naphthalene. The *exo*-cyclization mode could be achieved in some cases and led to the formation of isoindoles. The feasibility of a tandem radical cyclization was also established. Applications employing this new radical cyclization are underway.

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Supporting Information Available: Experimental procedures as well as ¹H NMR spectral and characterizations. This material is available free of charge via the Internet at http://pubs.acs.org.

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⁽¹⁷⁾ Compounds 32 and 33 are not separable. Also found on occasions was the *central*-cyclized product as part of the inseparable mixture.

⁽¹⁸⁾ For a recent review on radical processes leading to bridged azabicycles, see: Sato, T.; Ikeda, M. *Heterocycles* **2003**, *59*, 429.